

Supplementary Materials for

On the potential of using dual-function hydrogels for brackish water desalination

Wael Ali ^{1,2,4}, Beate Gebert ², Sedakat Altinpinar ¹, Thomas Mayer-Gall ^{1,2},
Mathias Ulbricht ³, Jochen S. Gutmann*^{1,2} and Karlheinz Graf⁴

¹ Physikalische Chemie and CENIDE (Center for Nanointegration), Universität Duisburg-Essen,
Universitätsstr. 2, 45141 Essen, Germany

² Deutsches Textilforschungszentrum Nord-West gGmbH, Adlerstr. 1, 47798 Krefeld, Germany

³ Technische Chemie II and CENIDE (Center for Nanointegration), Universität Duisburg-Essen,
Universitätsstr. 5, 45141 Essen, Germany

⁴ Physikalische Chemie, Hochschule Niederrhein, Adlerstr. 32, 47798 Krefeld, Germany

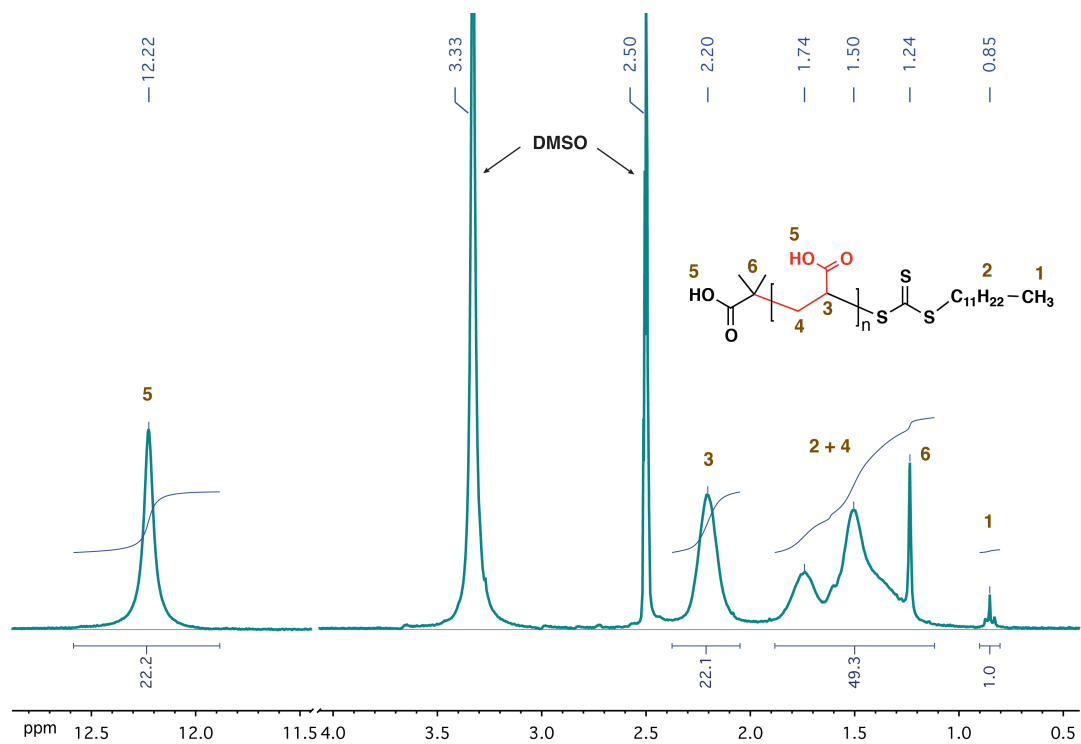


Figure S 1 Partial $^1\text{H-NMR}$ spectrum of poly(acrylic acid) macromolecular RAFT agent (PAA-TTC).

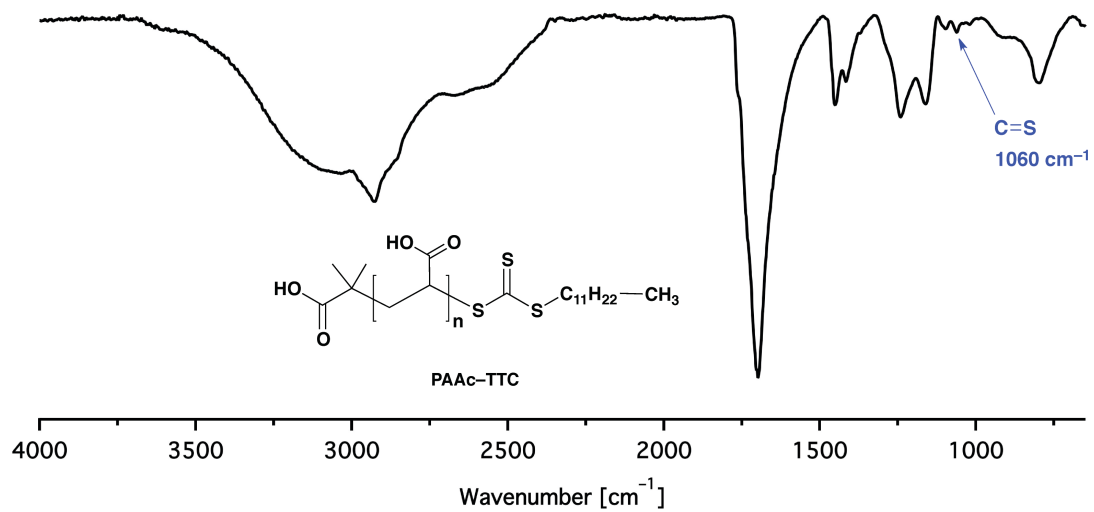


Figure S 2 FTIR spectrum of poly(acrylic acid) macromolecular RAFT agent (PAA-TTC).

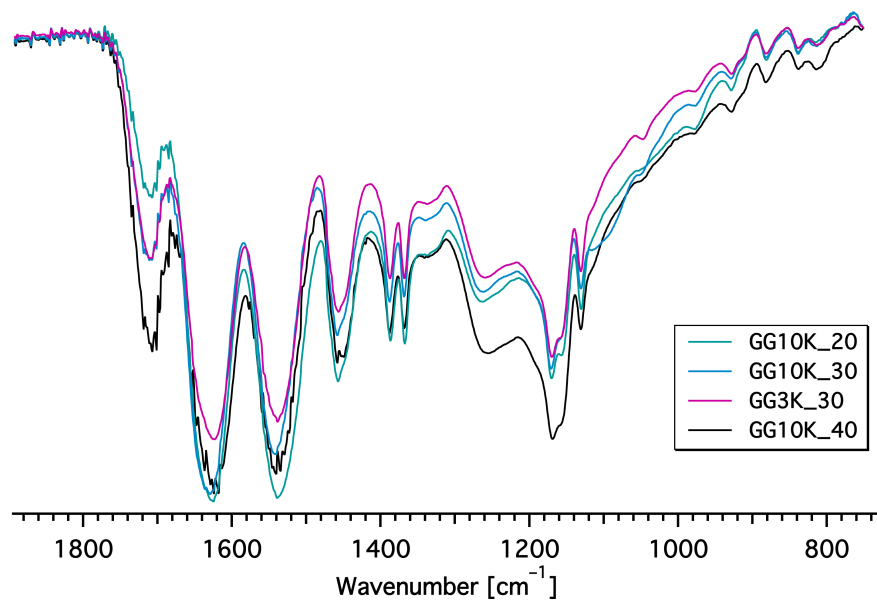
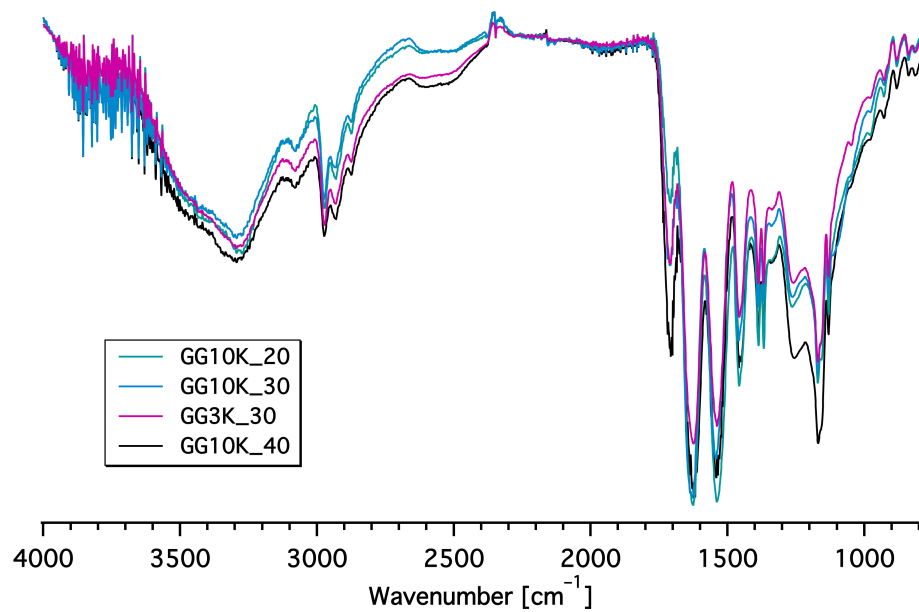


Figure S 3 FTIR spectra of hydrogels investigated in detail in this work.

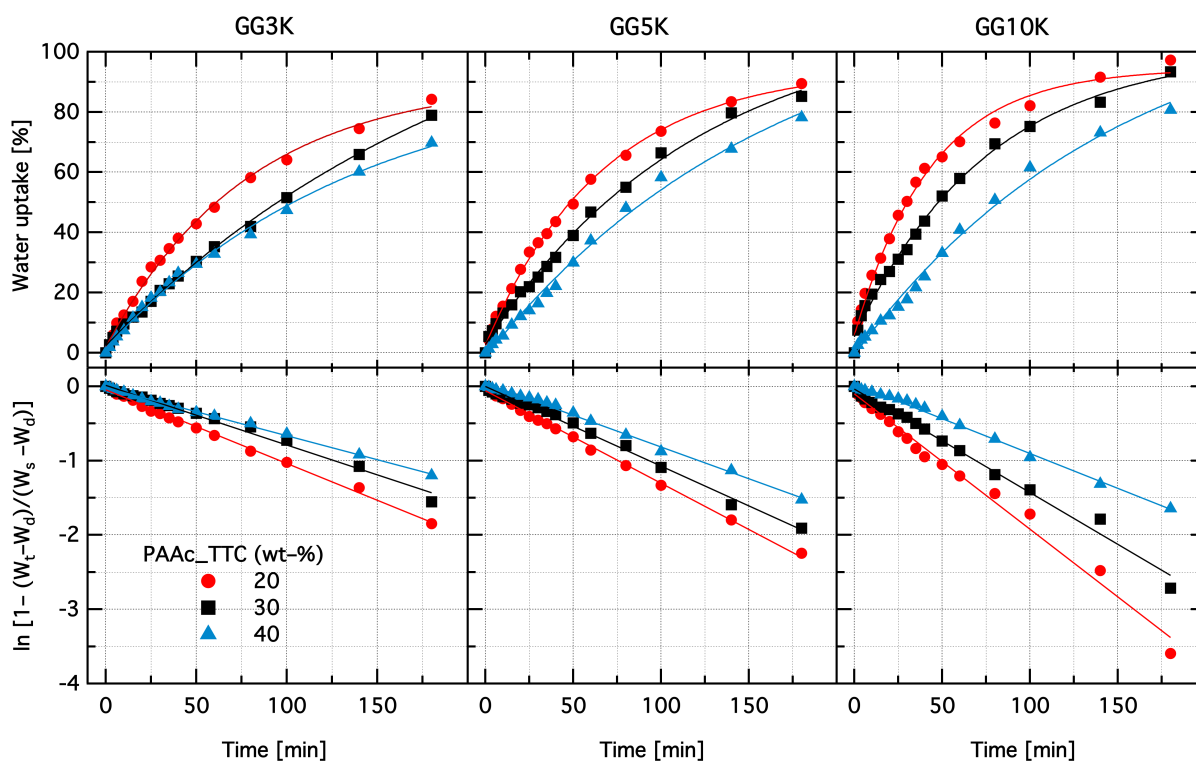


Figure S 4 Plots of water uptake as a function of swelling time in deionised water (DI) at room temperature (23 °C) (*upper plots*) and a semilogarithmic plots as first-order rate analysis was applied to the time dependence of the swelling (*lower plots*) to obtain the swelling rate constant, from which the half-swelling time ($t_{1/2}$) was obtained. The sample shape and size were constant for all hydrogels (cylinder, 5 mm in diameter and 2 mm in thickness).

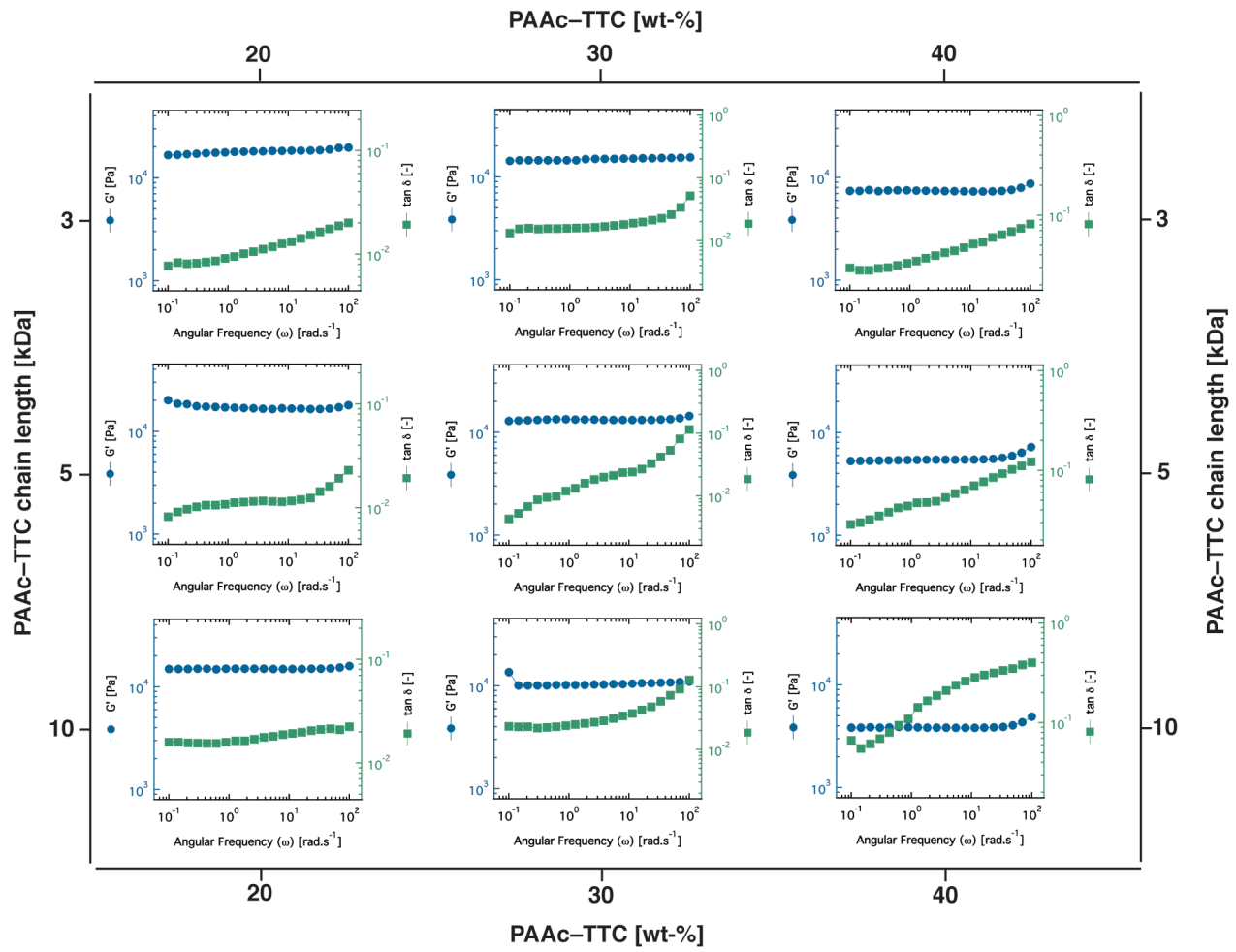


Figure S 5 Storage modulus (G') and loss tangent ($\tan \delta$) as a function of angular frequency (ω) of all hydrogels.

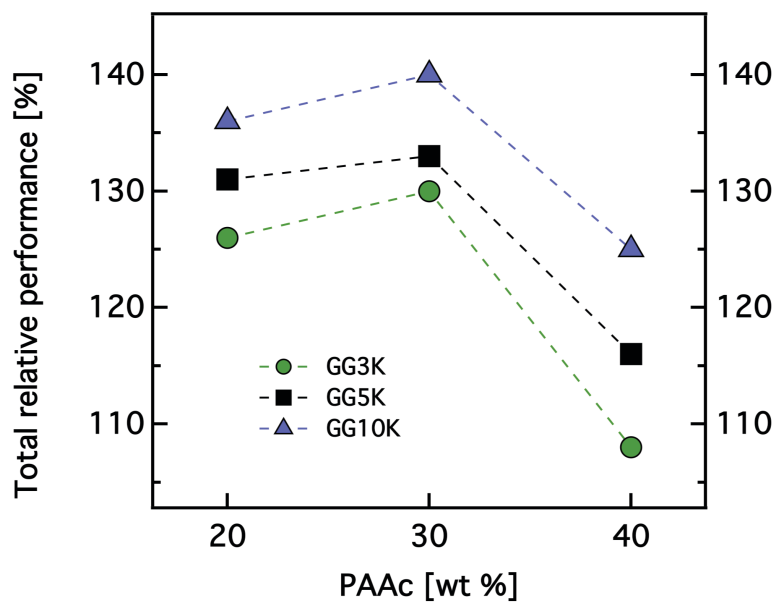


Figure S 6 The performance of hydrogels in terms of total relative performance (water recovery and salt rejection) as a function of grafted-PAAc content. The dashed lines are a guide to eye.